

Synthesis, Antimicrobial Activity Studies and Molecular Property Predictions of Schiff Bases Derived from ortho-Vanillin

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ABSTRACT

Schiff bases are known to possess anticancer, antibacterial, antifungal, antitubercular, anti-inflammatory, antibacterial and antimalarial properties. At the same time, in recent years, prediction of drug-likeness, molecular, absorption, distribution, metabolism, and excretion (ADME) properties using in silico techniques has become a standard procedure for the evaluation of clinical usable molecules. In this study, Schiff base structured 2-methoxy-6-[[2-alkyl/arylethyl]imino]methyl}phenol derivatives were synthesized from 3-methoxysalicylaldehyde (*o*-vanillin). The antibacterial and antifungal activities of these compounds were determined against *Staphylococcus aureus*, *Escherichia coli*, *Candida albicans* and *Pseudomonas aeruginosa* using agar-well diffusion and broth microdilution techniques. Further analysis was conducted using the in silico technique to predict the drug-likeness, molecular and ADME properties of these molecules. Among all the compounds, 2-[[4-fluorophenethyl]imino]methyl}-6-methoxyphenol (Compound 9) exhibited the highest activity with good minimum inhibition concentration and radius of inhibition zone values against *Candida albicans*.

Keywords: Schiff bases, Antifungal, *Candida albicans*, Molecular properties, In silico

INTRODUCTION

Schiff bases (azomethines) are compounds containing structures of R-CH=N-R' (R¹H), where R and R' can be either alkyl, aryl, cycloalkyl or heterocyclic groups¹⁻³. These compounds are very well recognized and due to their unusual properties, they are used in a wide range of application areas including organic

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and bioinorganic chemistry as common non-enzymatic/enzymatic intermediates, coordination and supramolecular chemistry as common ligands as well as biomedical applications and material sciences⁴⁻¹¹. They have also been reported to show antibacterial¹²⁻¹⁷, antifungal¹⁴⁻¹⁶ and antitumoral¹⁸⁻¹⁹ activities.

Schiff bases can be synthesized by nucleophilic addition followed by dehydration reaction cascade between aliphatic or aromatic amines and active carbonyl compounds. Various derivatives of these compounds formed from the reaction between salicylaldehydes and amines act as cellular growth regulators for plants²⁰ and have antibacterial²¹ and antimycotic²² activities. A wide range of different Schiff base ligands produced from *ortho*-hydroxyl substituted aromatic aldehydes have shown important coordinating properties for specific metal ions such as Al (III), Zn (II), Ag (II), Y (III), Co (II), Pb (II), Cu (II), Ni (II), Gd (III), Hg (II)²³⁻²⁸ and act as bidentate ligands for transitional metal ions²⁹⁻³³. At the same time, these compounds and/or their metal complexes have exhibited important anticancer and herbicidal properties³⁴⁻³⁵.

Similarly, a wide range of aldehyde and phenol structured secondary metabolites of plants isolated as crude extracts or oils have been studied for their potential antibacterial and antioxidant activities³⁶⁻³⁹. *Ortho*-vanillin (*o*-vanillin) can be considered one of the several examples of these compounds with antibacterial and antifungal activity^{37,40-41} as well as irritant properties.

In the course of our study, we synthesized 2-methoxy-6-[(2-alkyl/arylethyl)imino]methylphenol derivatives, Schiff bases of *ortho*-vanillin and 2-aryl/alkylethaneamines. The antibacterial and antifungal activities of the compounds were determined against *Staphylococcus aureus*, *Escherichia coli*, *Candida albicans* and *Pseudomonas aeruginosa* using agar-well diffusion and broth microdilution techniques. Furthermore, their absorption, distribution, metabolism, and excretion (ADME), drug-likeness and molecular properties were predicted using *in silico* techniques. Although Compounds 1, 3, 4, 7, 8, 13, 14 and 15 had been previously synthesized using different methods and investigated in other subjects⁴²⁻⁴⁹, in this study, we resynthesized them and performed a microbiological and prediction analysis to test their antibacterial and antifungal activities.

METHODOLOGY

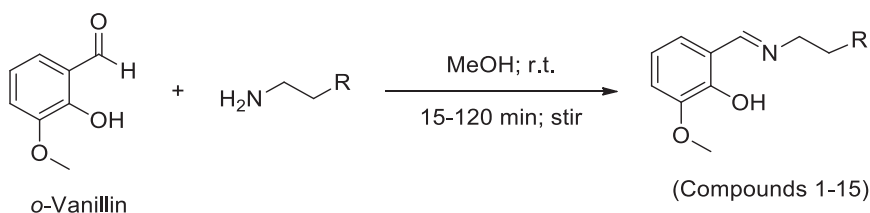
Chemistry

All chemicals were purchased from Aldrich Chemical Co. (Steinheim, Germany). Melting points were determined with a Mettler-Toledo FP62 capillary melting point apparatus (Columbus, OH, USA) and uncorrected. IR spectra (KBr) were recorded on a PerkinElmer Spectrum One FT-IR spectrometer (Waltham,

MA, USA) and $^1\text{H-NMR}$ spectra were obtained by Bruker DPX-400, 400 MHz High Performance Digital FT-NMR. All chemical shift values were recorded as δ (ppm). Mass spectra were recorded using an Agilent 1100 series LC/APCI/MS 1946 G spectrometer in the negative ionization mode. The purity of the compounds was checked by thin-layer chromatography on silica gel-coated aluminum sheets (Merck, 1.005554, silica gel HF254–361, Type 60, 0.25 mm; Darmstadt, Germany). The elemental analyses were performed with a Leco CHNS 932 analyzer (Leco Corp., MI, USA) and were found to be within $\pm 0.4\%$ of the theoretical values for C, H and N.

General synthesis for Schiff derivatives

Equimolar quantities (0.01 mol) of 3-methoxysalicylaldehyde and 2-aryl/alkylethaneamines were dissolved in methanol and stirred at room temperature for 10 to 120 min to obtain a clear solution using activated molecular sieves. Then, depending on the type of the substance, the solutions were either kept at room temperature or refrigerated overnight. The solutions that had precipitates were filtered, recrystallized, washed with cold methanol three times and dried in a vacuum desiccator. For the solutions that did not have any precipitates, column chromatography was performed using ethanol-ethyl acetate (1:1 v/v) as a mobile phase and the final compounds were obtained as either dark to light yellow solid crystals or yellow oils (Scheme 1).



Scheme 1: Synthetic pathway followed for the preparation of 2-methoxy-6-[(2-alkyl/arylethyl)imino]methylphenol derivatives (Compounds 1-15)

2-methoxy-6-([2-(pyridin-2-yl)ethyl]imino)methylphenol (Compound 1)

Yield 80%, yellow oil. IR (KBr) $\bar{\nu}_{\text{max}}$ (cm^{-1}): 3444 (s, OH), 3000 (CH, aromatic), 2795 (CH, aliphatic), 1632 (C=N), 1254. $^1\text{H-NMR}$ (400 MHz, DMSO- d_6 , δ): 3.2 (t, $J=7.10$ Hz, 2H, Py- CH_2), 3.90 (s, 3H, O- CH_3), 4.1 (t, $J=7.10$ Hz, 2H, $\text{CH}_2\text{-N=C}$), 5.30 (s, 1H, OH), 6.80-7.2 (m, 3H, Ph), 7.2-7.60, (m, 3H, Py), 8.2 (s, 1H, HC=N), 8.55 (d, 1H, Py). MS 256.1 (M^+). Anal. calcd for $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}_2$: C, 70.29; H, 6.29; N, 10.93. Found: C, 70.26; H, 6.31; N, 10.90.

**2-methoxy-6-({[2-(morpholin-4-yl)ethyl]imino}methyl)phenol
(Compound 2)**

Yield 70%, yellow crystals, mp 38-39 °C. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3008 (CH, aromatic), 2938 (CH, aliphatic), 1633 (C=N), 1253. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 2.37 (t, J=7.00 Hz, 4H, Mor), 3.2 (t, J=7.10 Hz, 2H, Mor-CH₂), 3.5 (t, J=7.00 Hz, 4H, Mor), 3.90 (s, 3H, O-CH₃), 4.0 (t, J=7.10 Hz, 2H, CH₂-N=C), 5.35 (s, 1H, OH), 6.90-7.1 (m, 3H, Ph), 8.25 (s, 1H, HC=N). MS 264.1 (M⁺). Anal. calcd for C₁₄H₂₀N₂O₃: C, 63.62; H, 7.63; N, 10.60. Found: C, 63.60; H, 7.65; N, 10.59.

**2-methoxy-6-({[2-(pyrrolidin-1-yl)ethyl]imino}methyl)phenol
(Compound 3)**

Yield 85%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3429 (s, OH), 2932 (CH, aromatic), 2794 (CH, aliphatic), 1632 (C=N), 1254. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 1.7 (t, J=6.5 Hz, 4H, Pyr H), 2.60 (t, J=6.5 Hz, 4H, Pyr H), 2.80 (t, J=7.10 Hz, 2H, CH₂-Pyr), 3.76 (t, J=7.10 Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.30 (s, 1H, OH), 6.40-6.80 (m, 3H, Ph), 8.1 (s, 1H, HC=N). MS 248.2 (M⁺). Anal. calcd for C₁₄H₂₀N₂O₂: C, 67.71; H, 8.12; N, 11.28. Found: C, 67.69; H, 8.15; N, 11.24.

**2-methoxy-6-({[2-(piperidin-1-yl)ethyl]imino}methyl)phenol
(Compound 4)**

Yield 75%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3429 (s, OH), 2932 (CH, aromatic), 2794 (CH, aliphatic), 1632 (C=N), 1254. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 1.6- 1.8 (m, 6H, Pip H), 2.60 (t, J=7.00 Hz, 4H, Pip H), 2.80 (t, J=7.10 Hz, 2H, CH₂-Pip), 3.76 (t, J=7.10 Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.35 (s, 1H, OH), 6.40-6.80 (m, 3H, Ph), 8.1 (s, 1H, HC=N). MS 262.2 (M⁺). Anal. calcd for C₁₅H₂₂N₂O₂: C, 68.67; H, 8.45; N, 10.68. Found: C, 68.69; H, 8.48; N, 10.72.

**2-{{[2-(2-chlorophenethyl)imino]methyl}-6-methoxyphenol
(Compound 5)**

Yield 82%, yellow crystals, mp 63-64 °C. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3010-3053 (CH, aromatic), 2856-2931 (CH, aliphatic), 1632 (C=N), 1253, 735-790. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 3.1 (t, J=7.10 Hz, 2H, CH₂-Ph), 3.80 (t, J=7.10 Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.35 (s, 1H, OH), 6.80-7.40 (m, 7H, Ph), 8.1 (s, 1H, HC=N). MS 289.1 (M⁺). Anal. calcd for C₁₆H₁₆ClNO₂: C, 66.32; H, 5.57; N, 4.83. Found: C, 66.27; H, 5.60; N, 4.80.

**2-{{[2,(4-dichlorophenethyl)imino]methyl}-6-methoxyphenol
(Compound 6)**

Yield 70%, yellow crystals, mp 103-104 °C. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3005-3089 (CH, aromatic), 2831-2961 (CH, aliphatic), 1633 (C=N), 1253, 702-710, 819-867. ¹H-

NMR (400 MHz, DMSO-*d*₆, δ): 3.1 (t, $J=7.10$ Hz, 2H, -CH₂-Ph), 3.84 (t, $J=7.10$ Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.30 (s, 1H, OH), 6.80-7.30 (m, 6H, Ph), 8.1 (s, 1H, HC=N). MS 323.1 (M⁺). Anal. calcd for C₁₆H₁₅Cl₂NO₂: C, 59.28; H, 4.66; N, 4.32. Found: C, 59.24; H, 4.69; N, 4.30.

**2-[(3-chlorophenethyl)imino]methyl}-6-methoxyphenol
(Compound 7)**

Yield 90%, yellow crystals, mp 39-40 °C. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3009-3060 (CH, aromatic), 2853-2936 (CH, aliphatic), 1633 (C=N), 1254, 780-731, 774-838. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 2.98 (t, $J=7.10$ Hz, 2H, CH₂-Ph), 3.80 (t, $J=7.10$ Hz, 2H, CH₂-N=C), 3.88 (s, 3H, O-CH₃), 5.30 (s, 1H, OH), 6.70-7.20 (m, 7H, Ph), 8.1 (s, 1H, HC=N). MS 289.1 (M⁺). Anal. calcd for C₁₆H₁₆ClNO₂: C, 66.32; H, 5.57; N, 4.83. Found: C, 66.28; H, 5.61; N, 4.81.

**2-[(2-fluorophenethyl)imino]methyl}-6-methoxyphenol
(Compound 8)**

Yield 83%, yellow crystal, mp 71-72 °C. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3010-3053 (CH, aromatic), 2856-2931 (CH, aliphatic), 1632 (C=N), 1253, 735-790. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 3.04 (t, $J=7.10$ Hz, 2H, CH₂-Ph), 3.84 (t, $J=7.10$ Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.30 (s, 1H, OH), 6.80-7.20 (m, 7H, Ph), 8.1 (s, 1H, HC=N). MS 273.1 (M⁺). Anal. calcd for C₁₆H₁₆FNO₂: C, 70.31; H, 5.90; N, 5.12. Found: C, 70.28; H, 5.93; N, 5.09.

**2-[(4-fluorophenethyl)imino]methyl}-6-methoxyphenol
(Compound 9)**

Yield 70%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3427 (s, OH), 2941-3058 (CH, aromatic), 2771-2827 (Al. CH), 1632 (C=N), 1254, 736-780. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 2.64 (t, $J=7.10$ Hz, 2H, CH₂-Ph), 3.70 (t, $J=7.10$ Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.30 (s, 1H, OH), 6.80-7.10 (m, 7H, Ph), 8.1 (s, 1H, HC=N). MS 273.1 (M⁺). Anal. calcd for C₁₆H₁₆FNO₂: C, 70.31; H, 5.90; N, 5.12. Found: C, 70.29; H, 5.94; N, 5.10.

**2-([2-(dimethylamino)ethyl]imino)methyl}-6-methoxyphenol
(Compound 10)**

Yield 70%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3433 (s, OH), 2941 (CH, aromatic), 2772-2821 (CH, aliphatic), 1633 (C=N), 1254, 736-780. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 2.3 (s, 6H, CH₃-N), 3.70 (t, $J=7.10$ Hz, 2H, CH₂-N.), 3.80 (t, $J=7.10$ Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.35 (s, 1H, OH), 6.80-7.10 (m, 3H, Ph), 8.1 (s, 1H, HC=N). MS 222.0 (M⁺). Anal. calcd for C₁₂H₁₈N₂O₂: C, 64.84; H, 8.16; N, 12.60. Found: C, 64.81; H, 8.19; N, 12.58.

2-([2-(diethylamino)ethyl]imino)methyl)-6-methoxyphenol (Compound 11)

Yield 88%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3435 (s, OH), 2968 (CH, aliphatic), 1631 (C=N), 1253. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 1.00 (t, J=8 Hz, 6H, CH₃-CH₂-N-), 2.40 (q, J=8 Hz, 4H, CH₃-CH₂-N-), 2.72 (t, J=7.10 Hz, 2H, -CH₂-N-), 3.65 (t, J=7.10 Hz, 2H, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.30 (s, 1H, OH), 6.40-7.10 (m, 3H, Ph), 8.1 (s, 1H, HC=N). MS 250.2 (M⁺). Anal. calcd for C₁₄H₂₂N₂O₂: C, 67.17; H, 8.86; N, 11.19. Found: C, 67.14; H, 8.89; N, 11.18.

2-([2-(2-chloroethyl)imino]methyl)-6-methoxyphenol (Compound 12)

Yield 88%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3418 (s, OH), 2965 (CH, aliphatic), 1651 (C=N), 1256. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 3.60 (t, 2H, -CH₂-Cl), 3.83 (t, 2H, -CH₂-N=C), 3.90 (s, 3H, OCH₃), 5.30 (s, 1H, OH), 6.50-7.10 (m, 3H, Ph), 8.1 (s, 1H, HC=N). MS 213.1 (M⁺). Anal. calcd for C₁₀H₁₂ClNO₂: C, 56.21; H, 5.66; N, 6.56. Found: C, 56.18; H, 5.70; N, 6.54.

2-methoxy-6-([2-(piperazin-1-yl)ethyl]imino)methyl)phenol (Compound 13)

Yield 75%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3427 (s, OH), 2938 (CH, aromatic), 2830 (CH, aliphatic), 1633 (C=N), 1254. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 2.0 (s, 1H, Ppz NH), 2.48 (t, 4H, Ppz H), 2.65 (t, 4H, Ppz H), 2.72 (t, 2H, CH₂-Ppz), 3.83 (t, 2H, J=7.10 Hz, CH₂-N=C), 3.90 (s, 3H, O-CH₃), 5.35 (s, 1H, OH), 6.50-7.90 (m, 3H, Ph), 8.1 (s, 1H, HC=N). MS 263.2 (M⁺). Anal. calcd for C₁₄H₂₁N₃O₂: C, 63.85; H, 8.04; N, 15.96. Found: C, 63.82; H, 8.07; N, 15.93.

2-([2-(1H-imidazol-2-yl)ethyl]imino)methyl)-6-methoxyphenol (Compound 14)

Yield 74%, yellow oil. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3437 (s, OH), 3005-3089 (CH, aromatic), 2831-2961 (CH, aliphatic), 1635 (C=N), 1254. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 2.90 (t, J=7.10 Hz, 2H, -CH₂-Imd), 3.83 (t, 2H, J=7.10 Hz, CH₂-N=C), 3.90 (s, 3H, OCH₃), 5.35 (s, 1H, OH), 6.60-7.30 (m, 5H, Ph and Imd), 8.1 (s, 1H, HC=N), 13.00 (1H, Imd. NH). MS 245.1 (M⁺). Anal. calcd for C₁₃H₁₅N₃O₂: C, 63.66; H, 6.16; N, 17.13. Found: C, 63.63; H, 6.19; N, 17.10.

2-methoxy-6-[(phenethylimino)methyl]phenol (Compound 15)

Yield 88%, yellow crystal, mp 76-77 °C. IR (KBr) $\bar{\nu}_{\max}$ (cm⁻¹): 3023-3085 (CH, aromatic), 2859-2996 (CH, aliphatic), 1627 (C=N), 1253. ¹H-NMR (400 MHz, DMSO-*d*₆, δ): 3.0 (t, J=7.10 Hz, 2H, CH₂-Ph), 3.84 (t, J=7.10 Hz, 2H, CH₂-N=C), 3.90 (s, 3H, OCH₃), 5.30 (s, 1H, OH), 6.80-7.40 (m, 8H, Ph), 8.1 (s, 1H, HC=N). MS 255.1 (M⁺). Anal. calcd for C₁₆H₁₇NO₂: C, 75.27; H, 6.71; N, 5.49. Found: C, 75.25; H, 6.74; N, 5.46.

Microbiological Screening

The following test microorganisms were obtained from LGC Standards GmbH (Wesel, Germany): *Staphylococcus aureus* (*S. aureus*) ATCC 25923, *Escherichia coli* (*E. coli*) ATCC 25922, *Pseudomonas aeruginosa* (*P. aeruginosa*) ATCC 27853, and *Candida albicans* (*C. albicans*) ATCC 60193. All the synthesized compounds were dissolved in dimethyl sulphoxide (DMSO) to prepare a stock solution at 10 mg/mL.

Agar-well diffusion method

An adapted simple susceptibility screening based on agar-well diffusion was used⁵⁰⁻⁵². Each bacterium was suspended in a Mueller-Hinton broth (MHB) (Difco, Detroit, MI, USA) and the fungal sample, *C. albicans*, was suspended in a Sabouraud Agar Modified medium (BD 274720) for pre-incubation at 35±2 °C. Then, all the microorganisms were diluted until they approximately matched the turbidity of 0.5 McFarland standard ($1-2 \times 10^8$ cfu/mL). Later, they were flood-inoculated onto the surface of Mueller-Hinton (MH) and Sabouraud dextrose (SD) agars and dried. Five-millimeter diameter wells were cut from the agar using a sterile cork borer, and 20 µL of the stock substances was delivered into the wells. The plates were incubated for 16-24 h at 35±2 °C. Antibacterial activity was evaluated by measuring the radius of the inhibition zone against the test organism using a digital caliper. *o*-vanillin (10 mg/mL), ampicillin (10 µg/mL) and fluconazole (5 µg/mL) were used as standards. DMSO diluted to 1:10 was used as the solvent control to ensure that it did not have any effect on bacterial growth.

Broth microdilution method

The minimal inhibition concentration (MIC) values (µg/mL) for the organisms were determined using the methods recommended by the Clinical and Laboratory Standards Institute (CLSI) guidelines⁵³⁻⁵⁴. The antimicrobial effects of the substances against the *S. aureus*, *E. coli* and *C. albicans* microorganisms were tested quantitatively in broth media using double dilution. The broth microdilution method was not performed on *P. aeruginosa* since none of the compounds showed inhibition zones with the agar well diffusion method. The antibacterial and antifungal assays were performed in an MH broth (Difco) at pH 7.3 and a buffered yeast nitrogen base (Difco) at pH 7.0, respectively. MIC was defined as the lowest concentration with no bacterial or fungal growth. Ampicillin (10 µg/mL) and fluconazole (10 µg/mL) were prepared as stocks, then diluted in a range from 10 to 0.5 µg/mL using DMSO and tested as standard antibacterial and antifungal drugs, respectively. The tested dilutions ranged from 128 to 0.5 µg/mL using DMSO as the solvent for all compounds. The control samples prepared with the amounts of DMSO used in the dilutions did not show any inhibitory activity under these conditions.

Prediction of drug-likeness, molecular and ADME properties

All the molecules were prepared in 3D using the LigPrep module of Maestro (Schrodinger Inc.). ADME properties (46 molecular descriptors) were calculated using the QikProp program (Schrödinger 2015-3) in the normal mode. QikProp generates physically relevant descriptors and uses them to perform ADME predictions. An overall ADME-compliance score, the drug-likeness parameter (indicated by #stars), was used to assess the pharmacokinetic profiles of the compounds. The #stars parameter (ranging from 0 to 5) indicates the number of property descriptors computed by QikProp that fall outside the optimum range of values for 95% of known drugs.

The predicted descriptors were: central nervous system (CNS) activity (−2 for inactive to +2 for active); octanol/water partition coefficient, $\log P_{o/w}$ (−2.0 to 6.5); IC_{50} value for the block of HERG K^+ channels, $\log HERG$ (concern < −5); Caco-2 cell membrane permeability in $nm\ s^{-1}$, PCaco (: < 5 low to > 100 high);

Table 1: Antimicrobial activity of compounds using the microdilution (MIC, $\mu g/mL$) and agar well diffusion methods (diameter of inhibition zones in mm)^a

Minimal inhibition concentration ($\mu g/mL$) and diameter of inhibition zones (mm)

| Compound | <i>S. aureus</i> | <i>E. coli</i> | <i>C. albicans</i> | <i>P. aeruginosa</i> |
|--------------------|------------------|----------------|--------------------|----------------------|
| 1 | >128 (0) | >128 (1) | 16 (17) | (0) |
| 2 | >128 (0) | >128 (2) | 16 (17) | (0) |
| 3 | 32 (11) | >128 (0) | 16. (17) | (0) |
| 4 | 16. (15) | 32. (7) | 64. (22) | (0) |
| 5 | >128 (0) | >128 (0) | 64 (14) | (0) |
| 6 | >128 (0) | >128. (0) | >128. (14) | (0) |
| 7 | >128 (0) | >128 (3) | >128 (14) | (0) |
| 8 | >128 (0) | >128 (3) | 16 (21) | (0) |
| 9 | 32 (13) | 16 (9) | 8. (25) | (0) |
| 10 | 32. (11) | 64 (4) | 16 (21) | (0) |
| 11 | 32 (11) | >128 (2) | 32 (17) | (0) |
| 12 | >128 (0) | 64 (5) | 16 (22) | (0) |
| 13 | >128 (0) | >128 (1) | 32 (19) | (0) |
| 14 | >128 (0) | >128 (3) | 16 (22) | (0) |
| 15 | >128 (0) | >128 (3) | 16 (19) | (0) |
| o-Vanillin | >128 (0) | 16 (8) | 16 (20) | (7) |
| Ampicillin | 2 (35) | >8 (10) | - | (18) |
| Fluconazole | - | - | 8 (25) | - |

^aThe results obtained from the agar well diffusion method are given in parentheses.

logarithm of predicted blood/brain barrier partition coefficient, $\log B/B$ (-3.0 to 1.0); apparent Madin-Darby canine kidney cell permeability (PMDCK) that mimic the blood-brain barrier for non-active transport in nm s^{-1} , PMDCK (< 25 poor to > 500 great); skin permeability, $\log K_p$ (-8.0 to -1.0); logarithm of binding constant to human serum albumin, $\log K_{\text{HAS}}$ (-1.5 to 1.2); qualitative human oral absorption, HOA (1: low, 2: medium, 3: high); percent of HOA (>80%: high, <25%: poor) (Table 2).

Table 2: Calculated drug-likeness, molecular properties and ADME predictions for Compounds 1-15 using QikProp

| Name | #stars | CNS | $\log Po/w$ | $\log HERG$ | PCaco | $\log BB$ | PMDCK | $\log Kp$ | $\log K_{hsa}$ | HOA ^a | %HOA |
|------|--------|-----|-------------|-------------|---------|-----------|----------|-----------|----------------|------------------|--------|
| 1 | 0 | 0 | 3.25 | -5.62 | 2723.09 | -0.42 | 1460.83 | -0.86 | 0.02 | 3.00 | 100.00 |
| 2 | 0 | 1 | 1.60 | -5.35 | 841.86 | 0.08 | 454.39 | -3.35 | -0.45 | 3.00 | 88.66 |
| 3 | 0 | 1 | 2.30 | -5.49 | 771.51 | 0.04 | 413.49 | -3.43 | -0.15 | 3.00 | 92.10 |
| 4 | 0 | 1 | 2.56 | -5.70 | 863.10 | 0.07 | 466.79 | -3.33 | -0.01 | 3.00 | 94.48 |
| 5 | 1 | 0 | 4.19 | -5.46 | 3553.49 | -0.18 | 3632.38 | -0.66 | 0.32 | 3.00 | 100.00 |
| 6 | 0 | 1 | 4.67 | -5.54 | 4079.69 | 0.05 | 10000.00 | -0.68 | 0.45 | 3.00 | 100.00 |
| 7 | 0 | 0 | 4.30 | -5.60 | 3423.27 | -0.15 | 4640.13 | -0.76 | 0.35 | 3.00 | 100.00 |
| 8 | 1 | 0 | 4.23 | -5.53 | 3421.94 | -0.24 | 2720.15 | -0.67 | 0.26 | 3.00 | 100.00 |
| 9 | 0 | 0 | 4.26 | -5.55 | 3438.21 | -0.20 | 3397.39 | -0.72 | 0.27 | 3.00 | 100.00 |
| 10 | 0 | 1 | 1.74 | -5.21 | 753.41 | 0.05 | 403.01 | -3.45 | -0.35 | 3.00 | 88.62 |
| 11 | 0 | 1 | 2.74 | -5.42 | 845.05 | -0.04 | 456.25 | -3.16 | -0.17 | 3.00 | 95.37 |
| 12 | 0 | 1 | 2.61 | -4.14 | 3405.88 | 0.02 | 4692.47 | -1.47 | -0.19 | 3.00 | 100.00 |
| 13 | 0 | 1 | 0.98 | -6.33 | 110.12 | 0.18 | 55.78 | -5.96 | -0.33 | 2.00 | 69.22 |
| 14 | 0 | -1 | 2.38 | -5.13 | 1254.71 | -0.73 | 632.21 | -1.74 | -0.24 | 3.00 | 96.34 |
| 15 | 0 | 0 | 4.02 | -5.72 | 3376.93 | -0.33 | 1843.40 | -0.61 | 0.25 | 3.00 | 100.00 |

^aThe assessment uses a knowledge-based set of rules, including checking for suitable values of percent human oral absorption, number of metabolites, number of rotatable bonds, $\log P$, solubility and cell permeability.

RESULTS AND DISCUSSION

Using the conventional method, the proposed Schiff bases of *o*-vanillin were successfully synthesized giving yields between 70-90%. In the IR spectra, all the compounds had a strong C=N stretching band at 1627–1651 cm^{-1} . The ¹H-NMR spectra of the compounds showed that the protons belonging to the compounds in the HC=N group were exhibited at δ 8.10–8.25 ppm as a singlet, in which both the IR ¹H-NMR data were accepted as evidence for the formation of an imine

bond. All the other protons were observed according to the expected chemical shift and integral values. The molecular ion peaks (M^+) of the compounds were examined under electron ionization and they confirmed the molecular weights of the compounds. MIC and inhibition zones of Compounds 1-15 were evaluated for their antibacterial and antifungal activities using *o*-vanillin, ampicillin and fluconazole as a standard for the microorganisms (*S. aureus*, *E. coli*, *C. albicans* and *P. aeruginosa*). Throughout the experiments, the parent compound *o*-vanillin was observed to be active for *E. coli* and *C. albicans* but there was minimum or no activity for *S. Aureus* and *P. aeruginosa*. The best activity of *o*-vanillin was seen against *C. albicans* with the MIC value of 16 $\mu\text{g/mL}$. Since none of the synthesized compounds was found to be active against *P. aeruginosa* during the measurement of the diameter for inhibition zones, further analysis on the compounds was not performed for this microorganism and broth micro-dilution experiments were not undertaken to determine MIC. According to the results on the inhibition zones of *S. aureus*, Compounds 3, 4, 9, 10, 11 were more active compared to the parent compound *o*-vanillin; however, they were not as active as ampicillin. Similarly, for *E. coli*, Compound 9 showed similar activity to *o*-vanillin but performed worse than ampicillin. All the compounds had moderate to strong activity against *C. albicans* resulting in similar inhibition zones. In broth dilution experiments, Compounds 1, 2, 3, 8, 10, 12, 14, 15 were found to have an MIC of 16 $\mu\text{g/mL}$, which was similar to that of *o*-vanillin. Another significant result was that Compound 9 had an MIC value of 8 $\mu\text{g/mL}$ presenting the same anti-fungal activity as fluconazole.

In this study, the drug-likeness, molecular and ADME properties of all compounds were promising presenting a drug-like/lead-like profile according to their #stars rankings. The combinations of HOA values being mostly around 3 (except Compound 13), %HOA values ranging from 88.62 to 100%, all PCaco values being high and log K_p values varying between -0.61 and -5.96 indicate that these compounds can be effectively used in both oral and topical preparations. It is important to note that the binding of drugs to plasma proteins (such as human serum albumin, lipoprotein, glycoprotein, α , β and γ globulins) greatly reduces the quantity of the drug in general blood circulation. In other words, the less bound a drug is, the more efficiently it can traverse cell membranes or diffuse. Furthermore, the logarithm of the predicted binding constant to human serum albumin, $\log K_{\text{HSA}}$, values of the synthesized Schiff bases varied between -0.45 to +0.45, which showed that these derivatives might circulate freely and easily traverse cell membranes. The HERG K^+ channel, best known for its impact on the electrical activity of the heart by synchronizing beating, appears to be the molecular target responsible for the cardiac toxicity of a wide range of therapeu-

tic drugs⁵⁵. HERG has also been found associated with modulating the functions of certain cells of the nervous system and with establishing and maintaining cancer-like features in leukemic cells⁵⁶. Thus, HERG K⁺ channel blockers are potentially toxic and the predicted IC₅₀ values often provide reasonable predictions for the cardiac toxicity of drugs in the early stages of drug discovery⁵⁷. The logHERG values of the compounds predicted with the in silico method were between -4.14 and -6.33 with only Compound 13 exhibiting a tendency for the blockage of this channel. As a basic rule, polar drugs cannot easily penetrate blood–brain barrier (BBB). The blood/brain partition coefficient (log BB), PMDCK and logPo/w values are useful to determine the penetration capacity of a compound from BBB. The values predicted for these parameters of the synthesized compounds were within the ranges defined for 95% of drugs. Moreover, the predicted CNS value of the compounds was between -1 and +1 indicating mild to medium activity.

Although the prediction results must be checked with actual experiments, the activity studies showed that the Schiff bases derived from *o*-vanillin derivatives could be considered as potential antifungal agents. Furthermore, investigations showed that the Schiff bases synthesized from the starting point of *o*-vanillin can act as a monovalent bidentate ligand, which develop antibacterial properties by combining the azomethine nitrogen and phenolic oxygen atom⁵⁸. Based on this information, we plan to undertake further research to evaluate the metal complexes of these ligands and their antibacterial/antifungal properties.

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